

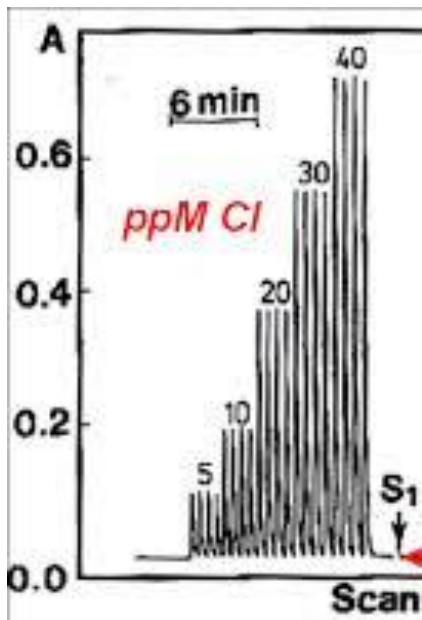
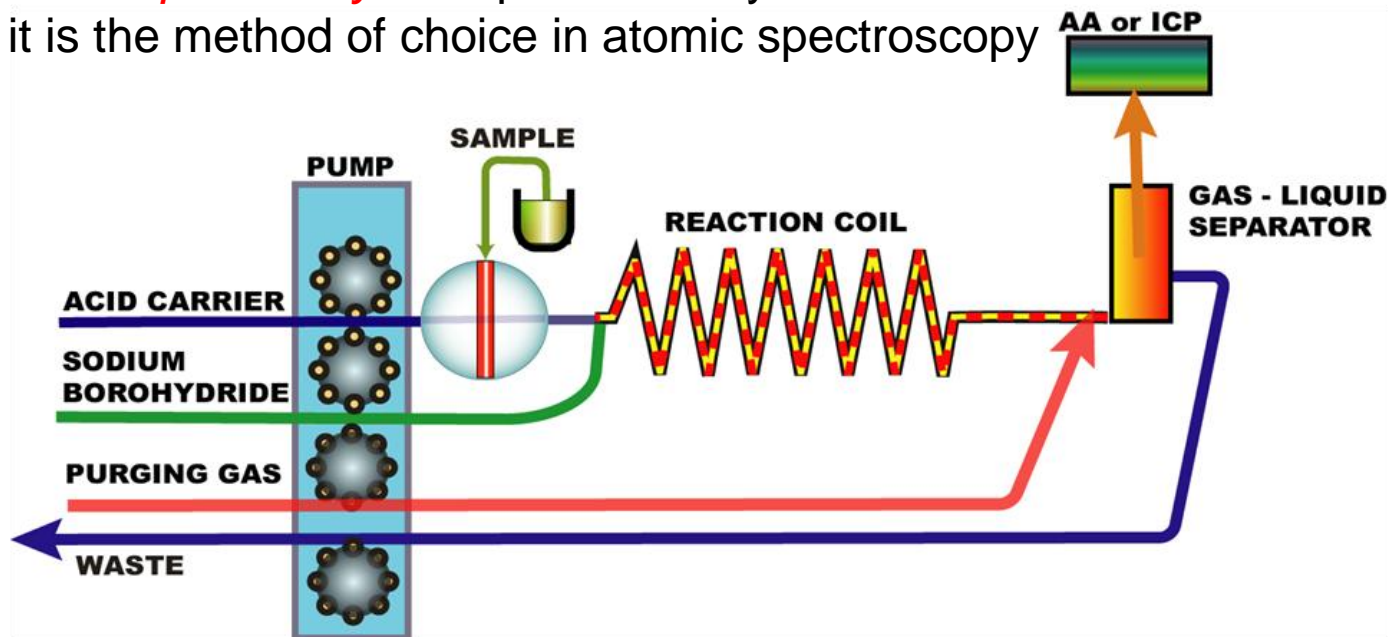
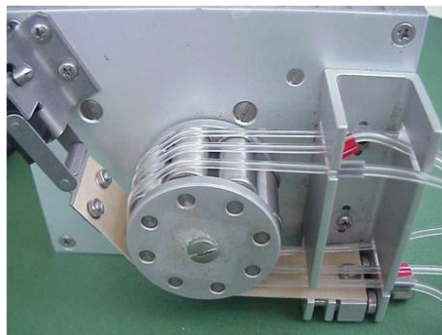
ATOMIC ABSORPTION ANALYTICAL TECHNIQUES

Sample Preparation:

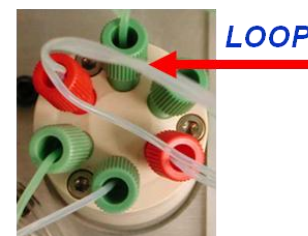
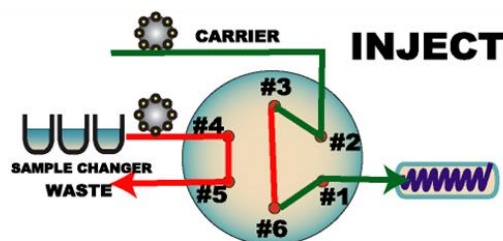
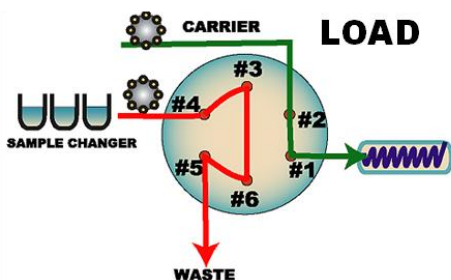
- A disadvantage of flame spectroscopic methods is the requirement that the sample be introduced into the excitation source in the form of a solution, most commonly an aqueous one.
- Unfortunately, many materials of interest, such as soils, animal tissues, plants petroleum products and minerals are not directly soluble in common solvents, and extensive preliminary treatment is often required to obtain a solution of the analyte in a form ready for atomization.
- Indeed, the decomposition and solution steps are often more time consuming and introduce more error than the spectroscopic measurement itself.
- Some of the common methods used for decomposing and dissolving samples for atomic absorption methods include treatment with hot mineral acids; oxidation with liquid reagents, such as sulfuric, nitric, or perchloric acids; combustion in an oxygen bomb or other closed container to avoid loss of analyte.

Sample Introduction by Flow Injection

- FIA is an automated method of chemical analysis in which a sample is injected into a flowing carrier solution that mixes with reagents before reaching a detector.
- when the highest level *of repeatability* is required or only *a small amount of sample is available*, it is the method of choice in atomic spectroscopy

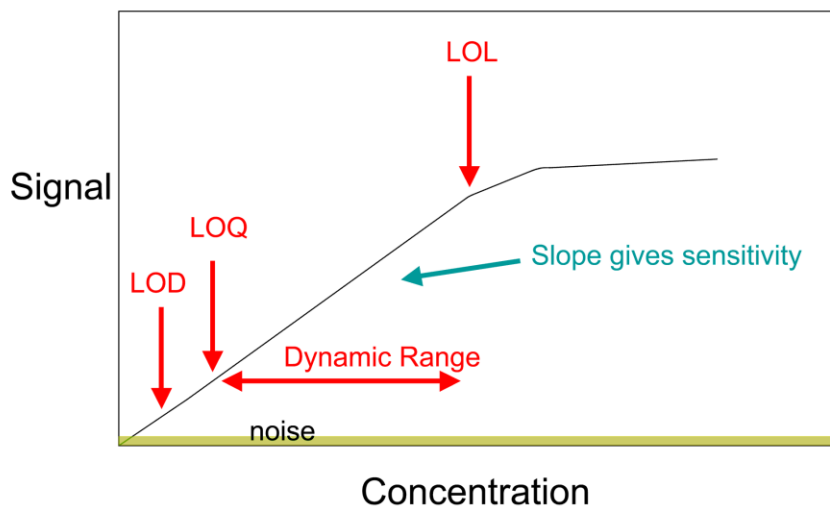
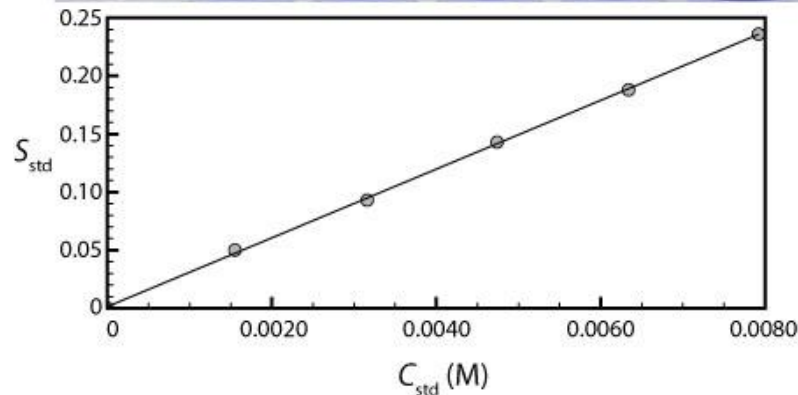


A two position, six port injection valve with a fixed loop for well defined sample volumes



Calibration Curve

- ✓ Atomic absorption should follow Beer's law, $A = \epsilon \cdot b \cdot C$
- ✓ however, departures from linearity are often encountered,
- ✓ A calibration curve that covers the range of concentrations found in the sample should be prepared.



➤ **LOD** is the lowest quantity of a substance that can be distinguished from the absence of that substance (a blank value) within a stated confidence limit

➤ LOD defined as $3 \times$ standard deviation of the blank, and at the LOQ defined as $10 \times$ standard deviation of the blank

A calibration curve plot showing limit of detection (LOD), limit of quantification (LOQ), dynamic range, and limit of linearity (LOL).

Standard Addition Method:

➤ It is widely used in atomic absorption spectroscopy in order to partially or completely compensate for the chemical and spectral interferences introduced by the sample *matrix*.

➤ *differing amounts of* standard is added directly to the same amount of aliquots of analyzed sample, diluted to a volume and analyzed at the wavelength of interest.

This process is also called spiking.

